Chemguide - questions

AMIDES: INTRODUCTION

- 1. a) Draw the structures for the following amides. In each case, you should show the bonding in the amide group, but you can simplify hydrocarbon groups as, for example, CH₃CH₂.
 - (i) ethanamide
 - (ii) propanamide
 - (iii) 2-methylpentanamide
 - (iv) N-methylbutanamide
- 2. a) Explain briefly why (apart from methanamide) the amides are low melting point solids.
 - b) Why are the small amides soluble in water?